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## Anthelmintic Activity of Newly Synthesized Moities of Fluoro Benzothiazole Schiff's Base.

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### ABSTRACT

4-Fluoro-3-chloroaniline treated with Potassium thiocyanate in presence of Glacial acetic acid and bromine was converted into 2-amino-6-fluoro-7-chlorobenzothiazole. The synthesized compound in presence of p-dimethylaminobenzaldehyde refluxed in ethanol to obtained 6-fluoro-7-chloro benzothiazole Schiff's base. The above said compound was treated with ortho, meta and para nitroanillines, ortho, meta, para chloroanillines, morpholino, Piperazine, diphenylamine in the presence of DMF to obtain different new moities. Some new moities showed promising anthelmintic activity.

**KEY WORDS:** Benzothiazole, Schiff's base, Anthelmintic Activity.

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## INTRODUCTION

Reports on fluorine[1] incorporated bioactive moieties are playing pharmacological activity, Amongst the heterocyclic compounds containing sulphur and nitrogen have attracted maximum attention as they are full of ramifications especially in industrial and biological applications. Fluorobenzothiazoles [2] are found to possess broad spectrum of pharmacological activity of clinical importance like anti-tubercular [3], anti-cancer [4, 5], anti-inflammatory [6], anti-microbial [7, 8] properties. This enthused us to prepare new moieties of fluorobenzothiazoles schiff's [9,10] base in hope of obtaining potent clinically useful biodynamic moieties. The said new moieties were screened for their anthelmintic activity according to method described in detail by Kailashraj and Kurup [11] *Perituma posthuma* (Earthworms obtained from Horticulture Department) of nearly equal size were selected for present work.

## MATERIAL AND METHODS

Purity of compounds was checked by TLC. Melting points were determined by open capillaries method and uncorrected. IR spectra (NaCl) are recorded on FTIR (Schimadzu-84005) spectrophotometer using nujol mull technique.

### First Step

#### Synthesis of 2-amino-6-fluoro-7-chloro-(1,3)benzothiazole [12]

To the glacial acetic acid (20ml) which is cooled below room temperature, 8gm (0.08mol) of potassium thiocyanate and 1.45g (0.01 mol) of fluorochloroaniline was added. The mixture was placed in freezing mixture of ice and salt, mechanically stirred while 1.6ml of bromine in 6ml of glacial acetic acid was added, from a dropping funnel at such a rate that the temperature never raised beyond room temperature. After all the bromine was added (105min), the solution was stirred for 2 hours below room temperature and at room temperature for 10 hours, it was then allowed to stand overnight, during which period an orange precipitate settle at the bottom, water (6ml) was added quickly and slurry was heated at 85<sup>o</sup>c on a steam bath and filtered hot. The orange residue was placed in a reaction flask and treated with 10ml of glacial acetic acid heated again to 85<sup>o</sup>c and filtered hot. The combined filtrate was cooled and neutralized with concentrated ammonia solution to p<sup>H</sup> 6. A dark yellow precipitate was collected. Recrystallised from benzene, ethanol of (1:1) after treatment with animal charcoal gave yellow plates of 2-amino-6-fluoro-7-chloro-(1,3) benzothiazole. After drying in a oven at 80<sup>o</sup>c, the dry material (1gm 51.02%) melted at 210-212<sup>o</sup>c. UV 307.4, 269nm, IR 1542cm<sup>-1</sup>(aromatic C=C) and 3475cm<sup>-1</sup> (NH<sub>2</sub>); 1456 cm<sup>-1</sup>(thiazole), 1215 cm<sup>-1</sup>(aromatic-F), 712 cm<sup>-1</sup>(aromatic-Cl).

## Second Step

### Synthesis of 2-[p-dimethylaminobenzylidene]-6-fluoro-7-chloro (1, 3) benzothiazole [12]

0.01 mol of 2-amino-6-fluoro-7-chloro (1, 3) benzothiazole with 0.015 mol solution of p-dimethylaminobenzaldehyde, added 20 ml ethanol and 3-4 drops of HCl and refluxed for 2-3 hrs. Solution cooled and poured into crushed ice. Recrystallised with benzene and ethanol.

### Preparation of various derivatives (E<sub>1</sub>-E<sub>9</sub>)

Schiff's base treated with equimolar quantities of various aromatic amines, refluxed for 2 hours in presence of DMF, recrystallised from alcohol and benzene.

Table No. 1 Analytical Data of the Compounds (E<sub>1</sub>-E<sub>9</sub>)










Compds	R	M.P (°C)	Yield (%)	Molecular Formula	Molecular Wt.	Elemental Analysis Data (Calculated in %)		
						C	H	N
E <sub>1</sub>		172	63	C <sub>22</sub> H <sub>18</sub> N <sub>5</sub> O <sub>2</sub> SF	435	60.68	4.13	16.09
E <sub>2</sub>		190	69	C <sub>22</sub> H <sub>18</sub> N <sub>5</sub> O <sub>2</sub> SF	435	60.68	4.13	16.09
E <sub>3</sub>		180	77	C <sub>22</sub> H <sub>18</sub> N <sub>5</sub> O <sub>2</sub> SF	435	60.68	4.13	16.09
E <sub>4</sub>		172	57	C <sub>22</sub> H <sub>18</sub> N <sub>4</sub> SFCl	425	62.11	4.23	13.17
E <sub>5</sub>		162	76	C <sub>22</sub> H <sub>18</sub> N <sub>4</sub> SFCl	425	62.11	4.23	13.17
E <sub>6</sub>		168	69	C <sub>22</sub> H <sub>18</sub> N <sub>4</sub> SFCl	425	62.11	4.23	13.17
E <sub>7</sub>		170	52	C <sub>20</sub> H <sub>17</sub> N <sub>4</sub> OSF	380	63.15	4.47	14.17
E <sub>8</sub>		167	58	C <sub>20</sub> H <sub>18</sub> N <sub>5</sub> SF	379	63.32	4.74	18.46
E <sub>9</sub>		132	77	C <sub>28</sub> H <sub>23</sub> N <sub>4</sub> SF	466	72.10	4.93	12.01

Table2. IR spectral assignments of synthesized compounds (E<sub>1</sub>-E<sub>9</sub>)

Compounds	Characteristic absorption bonds (in cm <sup>-1</sup> )						
	C=N Str.	Aro.C=C Str.	C-F Str.	NO <sub>2</sub>	C-Cl	3 <sup>o</sup> -Nitrogen	C-H
E <sub>1</sub>	1630	1688	1232	1396	---	3088	3477
E <sub>2</sub>	1570	1677	1243	1399	---	3084	3433
E <sub>3</sub>	1600	1667	1244	1409	---	3066	3409
E <sub>4</sub>	1630	1610	1228	---	1197	3080	3435
E <sub>5</sub>	1667	1600	1300	---	1198	3100	3400
E <sub>6</sub>	1700	1656	1345	---	1127	3090	3434
E <sub>7</sub>	1625	1693	1244	---	---	3074	3456
E <sub>8</sub>	1630	1606	1247	---	---	3095	3479
E <sub>9</sub>	1620	1780	1271	---	---	3074	3477

### Anthelmintic Activity Screening:

The synthesized compounds are screened for anthelmintic activity by using earthworms. Six earthworms of nearly equal size were placed in standard drug solution and test compounds solutions at room temperature. Normal saline was used as control. The standard drug and test compounds were dissolved in minimum quantity of DMF and the volume was adjusted upto 15 ml. with normal saline solution to get the concentration of 0.1%, 0.2% and 0.5%. Piperazine Citrate was used as a standard drug. The compounds were evaluated by the time taken for complete paralysis and death of earthworms. The mean lethal time for each test compound was recorded and compared with standard drug. The time taken by worms to become motionless was noted as paralysis time. To ascertain the death of the motionless worms, these were frequently applied the external stimuli, which stimulates and induces movement in the worms, if alive.

The mean lethal time and paralysis time of earthworms for different test compounds and standard drug are tabulated in Table 3.

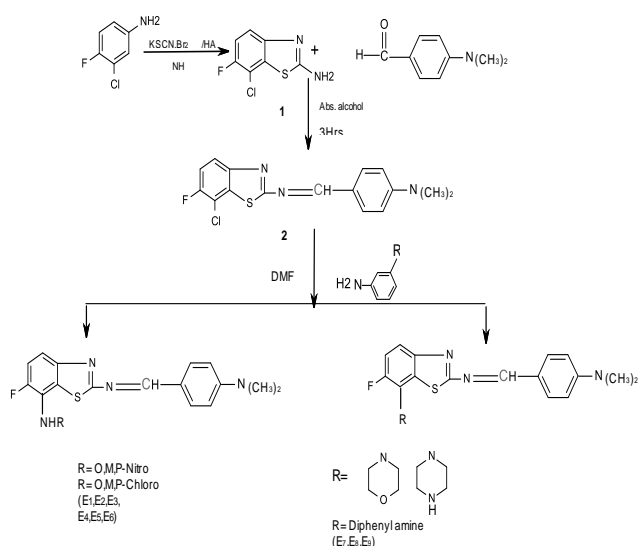
**Table3. Anthelmintic Activity of synthesized compounds (E<sub>1</sub>-E<sub>9</sub>)**

Compounds	Concentration (%)					
	Time for paralysis (min.)			Time for death (min.)		
Control	--	--	--	--	--	--
Piperazine citrate	65	53	44	75	60	55
E <sub>1</sub>	82	77	70	148	132	128
E <sub>2</sub>	79	74	68	142	134	126
E <sub>3</sub>	77	71	67	144	132	120
E <sub>4</sub>	78	73	65	141	129	119
E <sub>5</sub>	72	69	63	139	125	112
E <sub>6</sub>	75	65	58	146	133	122
E <sub>7</sub>	71	66	61	140	129	117
E <sub>8</sub>	69	60	57	128	122	114
E <sub>9</sub>	64	59	52	139	135	127

## RESULTS AND DISCUSSION

An attempt is made to synthesize the novel analogs of the fluorobenzothiazole schiff's base using *p*-dimethylamino benzaldehyde and screened for anthelmintic activity using *Perituma posthuma* (Earthworms obtained from Horticulture Department) of nearly equal size and some of the analogs showed significant activity.

### SCHEME





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